SHURBANOV, V1.

Vivid interest for chemistry and is application in everyday life. Khim i industrila 36 no.6:232 64.

SHURBANO, Vl., inzh.

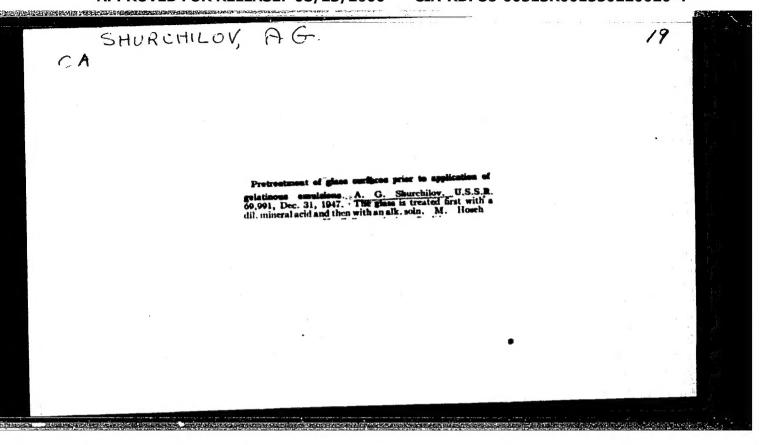
The nation-wide scientific technological conference on silicate industry. Khim i industriia 34 no.6:236-237 162.

SHURBANOV, V1., inzh.

A meeting on the exchange of experiences in the feed and tobacco industries. Tekh dele 501: 25 30 N 163.

SHURBANOV, VI.

Scientific technical conference at the Chemical Pharmaceutical Works, Sofia. Khim i industriia 36 no.10:393 '64.



"APPROVED FOR RELEASE: 08/23/2000

CIA-RDP86-00513R001550220020-4

PROTSENKO, P.I., SHURDUMOV, G.K.

Liquidus of the ternary reciprocal system consisting of rubidium and strontium nitrites and nitrates. Zhur.neorg.khim. 7 no.7:1644-1647 J1 :62. (MIRA 16:3)

1. Rostovskiy-na-Domi gosudarstvennyy universitet.
(Rubidium compounds) (Strontium compounds) (Systems (Chemistry))

PROTSENKO, P.I.; SHURDUMOV, G.K.

A PARTICIONAL DE LA SERIE MESANTA DE MANAGEMENTA DE LA SERIE DESERIE DE LA SERIE DE LA SER

Fusibility of the reciprocal system Cs, Sr II NO2, NO3. Izv.vys.ucheb. zav.;khim.i khim.tekh. 6 no.5:707-711 '63. (MIRA 16:12)

1. Rostovskiy-na-Donu gosudarstvennyy universitet, kafedra obshchey i neorganicheskoy khimii.

SHURDUMOV, G.K.; PROTSENKO, P.I.

Properties of solutions in the systems $NaNO_2 - Sr(NO_2)_2 - H_2O$ and $KNO_2 - Sr(NO_2)_2 - H_2O$ at 25°C. Zhur. neorg. khim. 9 no.5:1237-1241 My '64. (MIRA 17:9)

1. Rostovskiy gosudarstvennyy universitet.

PROTSENEO, P.1., SHUMDUMOV, G.K.

Differential thermal analysis of binary systems formed by alkali metal nitrites and strontium. Zhur. neorg. khim. 9 no.7:1692-1695 Jl *64. (MIRA 17:9)

1. Hostovskiy gosudarstvennyy universitet.

THE PROPERTY OF THE PROPERTY O

PROTSENKO, P.I.; SHURDUMOV, G.K.

Physicochemical properties of the systems $RbNO_2 = Sr(RO_2)_2 - H_2O$ and $GaNO_2 = Sr(NO_2)_2 - H_2O$ at 25°C. Thur. fiz. khim. 39 no.3:613-616 Mr 165. (MIRA 18:7)

1. Rostovskiy-na-Dona gosudarstvennyy universitet.

DECISHED, P.T. SHURPDANY, G.K.

outhility, electroconductivity, viscosity, and density in the stems $RbNO_2 = Sr(NO_2)_2 = H_2O$, $CsNO_2 = Sr(NO_2)_2 = H_2O$ at $25^{\circ}C_2$. Thur, neerg. khim, 10 no.2x480-484 F '65. (MIRA 18:11)

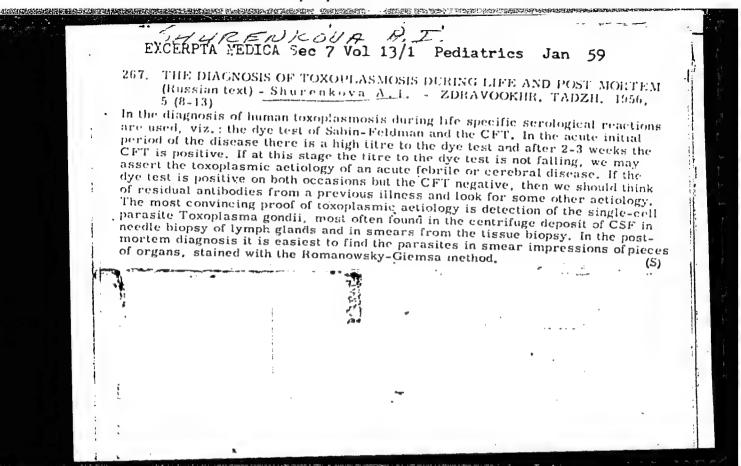
1. Rostovskiy-na-Donu gosudarstvennyy universitet. Submitted Sopt. 2, 1963.

LAGUNOV, V.; SHUREMOV, A.; TROFIMOV, M.; KOSTYKOV, I., slesar'; FERULEV, A.

In organizations of our society. Izobr.i rats. no.10: (MIRA 13:2)

1. Predsedatel' Yakutskogo oblastnogo soveta Vsesoyuznogo obshchestva izobretateley i ratsionalizatorov (for Ingunov).

2. Starshiy inzhener byuro tekhnicheskoy informatsii i izobretatel'-stva, L'vov (for Shuremov). 3. Predsedatel' soveta Vsesoyuznogo obshchestva izobretateley i ratsionalizatorov Vel'giyskoy obshchestva izobretateley i ratsionalizatorov, 4. Zavod bumazhnoy fabriki, g.Borovichi (for Trofimov). 4. Zavod "Soyuz," predsedatel' soveta Vsesoyuznogo obshchestva izobretateley i ratsionalizatorov, Leningrad (for Kostykov). 5. Predsedatel' zavodskoy organizatsii Vsesoyuznogo obshchestva izobretateley i ratsionalizatorov Iys'venskogo metallurgicheskogo zavoda, g.Iys'va, Permskoy oblasti (for Ferulev). (Efficiency, Industrial)



USSR / Zooparasitology. General Problems.

G-1

Abs Jour: Ref Zhur-Biol., No 20, 1958, 90988

; Shurenkova, A. I. Author

: Not given

: Certain Findings in the Study of Human Parasitic Inst Title

Diseases in Tadzhikstan

Orig Pub: Zdravokhr. Tadzhikistana, 1957, No 6, 23-30

Abstract: No abstract

Card 1/1

CIA-RDP86-00513R001550220020-4" APPROVED FOR RELEASE: 08/23/2000

KOMAROV, Ye.V.; PUSHLENKOV, M.F.; SHURENKOVA, M.Ye.

Factors determining the distribution of inorganic acids between aqueous and organic phases. Trudy Kom.anal.khim. 14:47-58 63. (MIRA 16:11)

SHURENOK, V. S.

"The Comparative Characteristics of the Clinical Manifestations of Typhoid Fever, Faratyphoid Fever A, and Faratyphoid B in Adults." Cand "ed Sci, L'vov Medical Inst, L'vov, 1955. (KL, No 15, Apr 55)

SO: Sum. No. 704, 2 Nov 55 - Survey of Scientific and Technical Dissertations Defended at USSR Higher Educational Institutions (16).

SHURGALTIA, P. P.

Growth - Study and Teaching

Teaching the subject "Physiological characteristics of the growing organism" (8th grade). Est. v shkole No. 2, 1952.

9. Monthly List of Russian Accessions, Library of Congress, July 1952. 1953, Uncl

AUTHOR:

Shurgal'skiy, E. F., Engineer

SOV/119-58-10-3/19

TITLE:

The Operation of Valves With Rubber Packings (Rabota klapanov

s rezinovym uplotneniyem)

PERIODICAL:

Priborostroyeniye, 1958; Nr 10; pp 7-11 (USSR)

ABSTRACT:

Air-crew masks, oxygen masks and pneumatic regulators must have valves that close perfectly. The conditions for a perfect operation of the valve packings are derived theoretically and experimentally or are investigated, respectively.

This investigation covers the two valve systems:

a) Valve with direct action, gas pressure "under the valve".

b) Nonreturn valve, gas pressure "on the valve".

Especially the sealing forces are derived. The correct operation of the valves is checked experimentally.

For the production of the valve seat an especially useful shape is given. The main attention has to be paid to the smoothness of the seat face (vvv 9 or vvvv 10). The relative pressure on the packing material which has to be applied to the unit of length of the valve seat to secure an absolute tightness is measured. (It is expressed by α in kg/cm). The

Card 1/2

The Operation of Valves With Rubber Packings

SOV/119-58-10-3/19

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a values are determined for the following diameters of the valve seats; 6.2; 4.2; 2.9 and 2.2 mm, with rubber of the type V-14 - 2 mm thick - (hardness according to Shoru = 80 - 82), and of the type N-14, group : G - 3.4 mm thick (hardness according to Shoru = 68 - 74) being used.

The dependence measured between the sealing force and the gas pressure to be controlled takes a linear course.

Also the increase of the sealing force was experimentally determined in the case that the sealing valve is tilted by 1 degree to the valve seat. From the graph obtained may be concluded that the increase of the sealing force is linearly proportional to the diameter of the valve seat and that it practically does not depend on the thickness and the hardness of the rubber. There are 7 figures and 7 references, 7 of which are Soviet.

Card 2/2

SHURGAL'SKIY, E.F.

Performance of the rubber and metal valves. Kauch. i rez. 22 no.8:14-18 Ag '63. (MIRA 16:10)

1. Moskovskiy institut khimicheskogo mashinostroyeniya.

SHURGANOV, Ya.

Useful practice. Avt. transp. 40 no.5:55 My '62. (MIRA 15:5)
(Transportation, Automotive)

KECHKER, M.I., SHURGAYA, Sh.I.

中国的国际中国的国际的国际的国际的国际的国际的国际的国际的国际的国际的国际的国际。 第二个国际国际中国

Study of the vectorcardiogram in healthy subjects. Klin.med. 36 no.8:87-93 Ag 158 (MIRA 11:9)

1. Iz pervoy kafedry terapii (zav. - deystvitel'nyy chlen AMN SSSR prof. M.S. Vovsi). TSentral'nogo instituta usovershenstvovaniya vrachey.

(VECTORCARDIOGRAPHY, in healthy subjects (Rus))

New 1541-type open-side boring and turning lathe. Stan. i instr.
28 no.11:19-21 N '57. (MIRA 10:12)
(Lathes)

PHASE I BOOK EXPLOITATION

SOV/4753

Safronovich, A.A., M.V. Bulatov, and A.M. Shurgin

- Karusel'nvve stanki; konstruktsiya i ekspluatatsiya (Vertical Boring Mills: Construction and Operation) Moscow, Mashgiz, 1960. 174 p. 8,000 copies printed.
- Managing Ed. for Literature on Metal Working and Machine-Tool Making (Mashgiz): V.I. Mitin, Engineer; Ed. of Publishing House: N.A. Ivanova; Tech. Ed.: V.D. El'kind.
- FURPOSE: This book is intended for technical personnel concerned with the operation of vertical boring mills at factories and for operators interested in improving their skills. It may also be used by students at schools of higher technical education as supplementary material for the course on "Metal-Cutting Machine Tools."
- COVERAGE: The authors describe the construction of Soviet and non-Soviet vertical boring mills. The book contains information on basic subassemblies, special constructional features, and modern constructions of these mills. Vertical boring mills are classified according to standard size and purpose. Trends in

Card

CIA-RDP86-00513R001550220020-4"

APPROVED FOR RELEASE: 08/23/2000

Vertical Boring Mills (Cont.)	SOV/4753
the construction of modern vertical mills are given for servicing and operating these mills There are 30 references, all Soviet.	
TABLE OF CONTENTS:	
The Purpose and Standard Sizes of Vertical Borin	g Mills 3
Description of the Constructions of Vertical Bor	ing Mills 5
Non-Soviet Vertical Boring Mills	28
Description of the Construction of Vertical Bori Subassemblies Frames for single-column mills Portal-type frames Beds and tables Cressrails Tool heads Speed drive	ng-Mill 32 34 35 51 58 75
Card 2/4	

BRW. B., V. Y.

WV-shaped (delta) wing on a smarrence flow.

The supersonic flow of gas past conical bodies representing a delta wing, the stabiliser of a missile, a fuselage wit, a delta-shaped wing and a fuselage with a stabiliser are investigated in this paper. The thickness of the body and with a stabiliser are investigated in this paper. The thickness of the body and the inclination angles relative to the incident flow are assumed as bein sufficiently small. Formulae are given determining the distribution of the pressure along the bodies and of the aerodynamic coefficients c ;c ,c ,c ,c .

In an appendix these coefficients are calculated. (First published in 1949).

Symposium of Theoretical Work on Aerodynamics, Oborongiz, 1957, 3,000 copies, Gentral Aero-Hydrodynamics Inst. imeni Prof. N. Ye. Zhukovskiy.

THE OWNER WHEN THE PROPERTY OF THE PROPERTY OF

CTT MCIR, 7. M.

"Flow past w res w th le lected allerons at the round species."

The flow past allerons of a supersonic gas flow is considered in presence of a wing located under a zero incidence angle. It is assumed that the thickness of the wing and of the allerons and the angles of deflection of the allerons are sufficiently small. Seven fundamental cases are considered depending on the shape of the wing and the allerons and on the Mach number of the incident flow. For each case formulae are given for calculating the pressure distribution (para. 2). In para. 3 the formulae are derived for determining the aerodynamic characteristics of the alleron for the first case. The results obtained in this paper can also be utilized for the aerodynamic calculation of other control or; and of aircraft. (First published in 1949).

Symposium of Theoretical Work on Aerodynamics, Oborongiz, 1957, 3,000 copies, Gentral Aero-Hydrodynamics Inst. imeni Prof. N. Ye. Zhukovskiy.

EWT(1)/EWG(v)/FCC/EEC-4/EEC(t)/EWA(h) Po-4/Pe-5/Pg-4/Pae-2/Peb/ UR/0293/65/003/002/0336/0340 ACCESSION NR: AP5009654 AUTHOR: Pletnev, V. D.; Shuridin, G. A.; Shalimov, V. P.; Shvachunov, I. N. TITLE: Dynamics of the geomagnetic trap and the origin of radiation belts SOURCE: Kosmicheskiye issledovaniya, v. 3, no. 2, 1965, 336-340 TOPIC TAGS: magnetosphere, solar wind, geomagnetic field, magnetic storm, force line, proton belt, electron belt ABSTRACT: The boundary of the magnetosphere created by the interaction between the solar wind and the geomagnetic field reaches a distance of 10 terrestrial radii on the day side of the earth. Electric currents on the boundary increase the magnetic field there. On the night side the magnetosphere is very extended. A particle may pass through the boundary of the magnetosphere because of a radial drift of the particle in an asymmetric magnetic field. The physical processes are studied in a magnetic field from parallele #70°. The regions permitting and prohibiting particle motion are determined,

ACCESSION NR: AP5009654

following Stormer's theory. Boundary currents diminish the magnetic field at neutral points. This effect shows up in the beginning of a magnetic storm. The combination of the current field and the dipole serves to straighten the force lines in the magnetosphere and stretch them towards the solar wind. The proton belt is nearer the earth than the electron belt. Orig. art. has: 3 figures and 2 formulas.

ASSOCIATION: none

SUBMITTED: 31Dec64

ENCL: 00

SUB CODE: AA, ES

NO REF SOV: 004

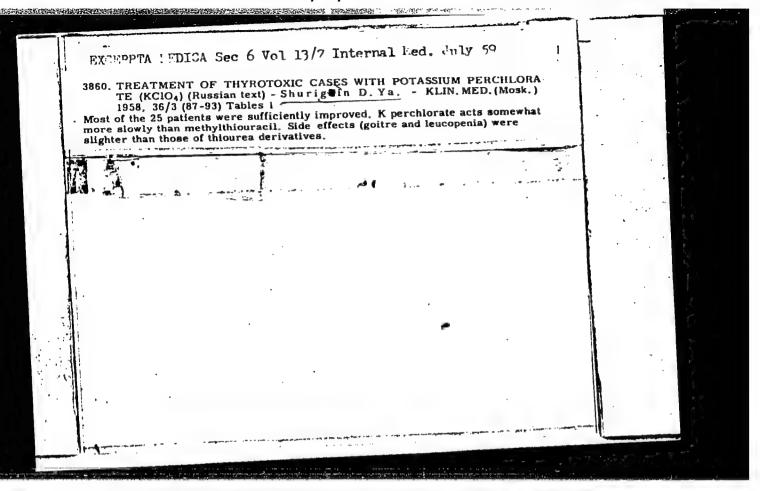
OTHER: 011

ATD PRESS: 3245

Card 2/2

KUDRIN, V.S.; KUDRINA, M.A.; SHURIGA, T.N.; GINZBURG, A.I., glavnyy red.; APEL'TSIN, F.R., zamestitel' glavnogo redaktora; CHERNYSHEVA, L.V., red.; HEUS, A.A., red.; GREKULOVA, L.A., red.; GRIGOR'YEV, V.M., red.; ZABOLOTNAYA, N.P., red.; MATIAS, V.V., red.; POKALOV, V.T., red.; RODIONOV, G.G., red.; STEPANOV, I.S., red.; CHERNOSVITOV, Yu.L., red.; SHMANENKOV, I.V., red.

[Rare-metal metasomatic formations associated with subalkaline granitoids.] Redkometal nye metasomaticheskie obrazovaniia, sviazannye s subshchelochnymi granitoidami. Moskva, Nedra, 1965. 145 p. (Geologiia mestorozhdenii redkikh elementov, no.25)



 SHURIK, R. E.

GUREVICH, Boris Samsonovich; MAKHOTINA, Nina Grigor'yevna; SHURIK, Rakhil Elyukomovna; BORISOVA, G.A., red.; SUDAK, D.M., tekhn. red.

[Fur articles, sheepskin coats, knit goods, sundries, perfumes and cosmetics; student manual for merchandise departments of institutes of Soviet commerce] Tovary: Pushno-mekhovye, ovchinno-shubnye, trikotashnye, galantereinye, parfiumerno-kosmeticheskie; uchebnoe posobie dlia tovarovednykh otdelenii tekhnikumov sovetskoi uchebnoe posobie dlia tovarovednykh otdelenii tekhnikumov sovetskoi torgovli. Moskva, Gos. izd-vo torg. lit-ry, 1957. 288 p. (Commercial products) (MIRA 11:7)

CIA-RDP86-00513R001550220020-4 "APPROVED FOR RELEASE: 08/23/2000

EWP(1) GS L 27887-65

AT5003952 ACCESSION NR:

\$/0000/64/000/000/0286/0298

AUTHOR: Malinovskiy, B. N.; Shurikhin, V. I.; Spynu, G. A.

TITLE: Digital method in the technology of design and manufacture of ship's hull

SOURCE: Nauchno-tekhnicheskoye obshchestvo priborostroitel'noy promyshlennosti. Nauchno-tekhnicheskoye soveshchaniye. 3d, Moscow, 1962. Vychislitel'naya tekhnika diya avtomatizatsii proizvodstva (Computer technology for the automation of production); trudy soveshchaniya. Moscow, Izd-vo Mashinostroyeniye, 1964, 286-298

TOPIC TAGS: program control, control computer, production planning, shipbuilding

ABSTRACT: The article describes a digital-computer-controlled automatic steelplate acetylene cutter for shipbuilding use, intended for automatization of laying out and cutting flat sheets of irregular shape. The heart of the control system is the transistorized UMShN (general purpose control computer) developed by Vychislitel nyy tsentr (Computation Center) AN UkrSSR. The program is recorded on the cutter motion program is recorded on the cutter motion program is recorded on the cutter motion program is recorded. The cutter motion program is 7 tracks of standard 35-mm perforated magnetic tape.

L 27887-65

ACCESSION NR: AT5003952

written in unitary code, and the technological commands as sinusoidal signals with approximate frequency 800 cps. The use of an interpolator makes it possible to move the cutter in steps of 0.2 mm. Some features of the interpolator are described. Some of the possibilities afforded by the introduction of computers into hull construction practice are discussed. It is pointed out in the conclusion that to take full advantage of programmed control, it is necessary to revise some of the existing hull design procedures. Orig. art. has: 5 figures, 3 formulas, and 3 tables.

ASSOCIATION: None

SUBMITTED: 01Sep64

ENCL: 00

SUB CODE: IE

NR REF SOV: 006

OTHER: 000

Card 2/2

SHURIKHINA, G. M.

Shurikhina, G. M. - "Exchange adsorption on cellulose materials," -- In table of contents third author: T. I. SHURIKHINA --- Materialy Tsentr. nauch.-issled. in-ta bumazh. prom-sti, Issue 37, 1948, p. 83-106 --- Bibliog: p. 104-06

So; U-3566, 15 Merch 53, (Letopis 'Zhurnal 'nykh Statey, No. 13, 1949)

THE DESTRUCTION OF THE ACTION OF THE PROPERTY OF THE PROPERTY

SOV/137-57-11-22236 D

Translation from: Referativnyy zhurnal, Metallurgiya, 1957, Nr 11, p 225 (USSR)

AUTHOR: Shurin, A.K.

然而公允允许是明显的证明的证明的证明的证明,在这个人只要的证明的证明的证明。

TITLE: Phase Diagrams of the Iron-Arsenic and the Iron-Carbon-

Arsenic Systems (Diagramma sistemy zhelezo-mysh'yak i

sistemy zhelezo-uglerod-mysh'yak)

ABSTRACT: Bibliographic entry on the Author's dissertation for the de-

gree of Candidate of Technical Sciences, presented to the In-t chernoy metallurgii AN USSR (Institute of Ferrous Metallurgy,

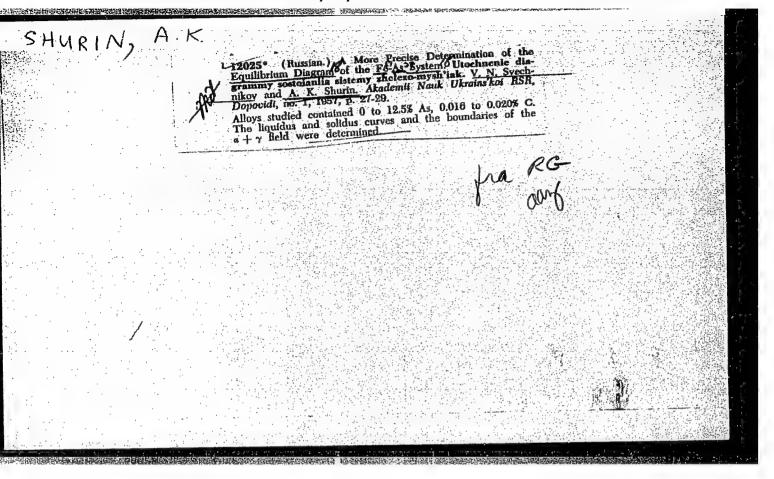
Academy of Sciences, UkrSSR) Dnepropretrovsk, 1957

ASSOCIATION: In-t chernoy metallurgii AN USSR (Institute of Ferrous

Metallurgy, UkrSSR), Dnepropetrousk

Card 1/1

"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001550220020-4



"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001550220020-4

SHURIN, A.K.; SVYECHNIKOV, V.M., akademik.

的引起**对自己的影响的影响的影响影响的影响的 1980年 1980年** 1980年 1980年

Investigation of the equilibrium diagram of an Fe-C-As system in the region of iron-rich alloys. Dop. AN URSR no. 2:138-140 157. (MLRA 10:5)

1. Akademiya nauk URSR (for Svyechnikov). 2. Institut metalofiziki AN URSR.

(Iron alloys--Metallography)

137-58-5-10520

Translation from Referativnyy zhurnal, Metallurgiya, 1958, Nr 5, p 234 (USSR)

AUTHORS. Svechnikov, V.N., Shurin, A.K.

DESCRIPTION OF THE PROPERTY OF

TITLE: The Iron-arsenic Phase Diagram (Diagramma sostoyaniy zhelezo-myshiyak)

PERIODICAL. Sb. nauchn. rabot In-ta metallofiz. AN UkrSSR, 1957 Nr 8, pp 51-57

ABSTRACT: The alloys were made in a Tammann furnace in a current of Ar with electrolytic Fe and an As alloying compound, the latter being produced by briquetting of 30% powdered As and 70% electrolytic Fe. The liquidus line was determined by thermal analysis in a Kurnakov pyrometer. Homogenizing of the alloys was performed at 800-1200°C. The investigation was by the study of the quenched microstructure. In alloys with Fe', As forms an enclosed rield. The concentrations of solid and rootions in equilibrium at 11500, in which the x loop is of maximum extent, were found to be 1.75 and 4.2% respectively. The eutectic transformation temperature was found to be 840°. The maximum solubility of As in QFe, determined by X-ray analysis of the structure is 12.4% and 840° and 5% at room temperature. Card 1/1

1. Arsenic-iron alloys--Phase studies

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137-58-6-13258

Translation from: Referativnyy zhurnal, Metallurgiya, 1958, Nr 6, p 298 (USSR)

AUTHORS: Shurin, A.K., Svechnikov, V.N.

TITLE: The Iron-carbon-arsenic Alloys (Sistema zhelezo-uglerod-

mysh'yak)

PERIODICAL: Sb. nauchn. rabot In-ta metallofiz. AN UkrSSR, 1957, Nr 8,

pp 58-64

ABSTRACT: The alloys were smelted of metallic As (96%), Armco Fe,

electrolytic Fe, and an alloy containing ~5% C in a Tammann furnace. Homogenization was done at 900-1200°C during 6-8 hr, which fully eliminated dendritic liquation of As. Investigation was conducted by the methods of differential dilatometric analysis and by microstructural methods of measuring of resistivity during heating. Four cross sections were constructed for constant As contents: 0.8, 1.4, 2.8, and 4.5%. As was above points A₁ and A₃. At 1.75% As the X+X and S+X regions unite. At 2.4% As the X and S regions unite. The

temperature of the solidus drops sharply with an increase in As content (by 220-320° depending on C content.) The temperature

Card 1/2 of the eutectic transformation L + Fe₃C with 4.5% As

137-58-6-13258

The Iron-carbon-arsenic Alloys

decreases to 960°. In Fe-C-As alloys As is fully dissolved in ferrite. A structural abnormality was detected: hypoeutectoid cast alloys contain more ferrite than annealed ones, while both cast and annealed hypereutectoid alloys contain structurally free ferrite.

A.V.

1. Arsenic-carbon-iron alloys--Properties 2. Arsenic-carbon-iron alloys--Microstructure 3. Arsenic-carbon-iron alloys--Temperature factors

Card 2/2

"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001550220020-4

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SVECHNIKOV, V.N.; KOCHERZHINSKIY, Yu.A.: PAN, V.M.; SHURIN, A.K.

Investigating chromium-niobium-vanadium alloys. Issl. po zharopr.
splav. 3:168-177 ' 58, (MIRA 11:11)
(Chromium-niobium-vanadium alloys—Metallography)
(Phase rule and equilibrium)

SOV/126-6-4-13/34

Syechnikov. V.N., AUTHOR:

SUPPLIER TO THE PROPERTY OF TH

Pan, V.M. Shurin, A.K.

Effect of Phosphorus and Arsenic on the Lattice Parameter TIPLE:

and Hardness of -Iron (Vliyaniye fosfora i mys / yaka

na parametr kristallicheskoy reshetki i twardost

al'fa-zheleza)

PERIODICAL: Fizika Metallov i Metallovedeniye, 1958, Vol. 6,

Nr 4, pp 662-664 (USSR)

High purity electrolytic iron was used for the ABSTRACT:

preparation of the experimental Fe-P and Fe-As alleys melted in vacuum (10-4 mm Hg) in the former, and in argon

in the latter case. The alloying elements were

introduced in the form of master alloys of the eutectic

composition (10.5% P or 30% As) prepared by powdered

metallurgy methods (sintered in evacuated quartz ampoules). No losses of the alloying elements occured

on smelting, and the carbon content of the Fe-P and Fe-As alloys was 0.004 - 0.010% and 0.016 - 0.020%

respectively. To remove the segregation effects the Card 1/4

SOV/126-6-13/34

Effect of Phosphorus and Arsenic on the Lattice Parameter and Hardness of & -Iron

alloys were subjected to a homogenising diffusion annealing treatment in the single (or 8) phase temperature region. The powder specimens for the X-ray analysis prepared by filing were annealed for 2 hrs at 800-1050°C, i.e. in the < -phase region, and cooled rapidly. For determining the macro and micro-hardness numbers, the Vickers hardness testing machine (5 or 20 kg load) and a PMT-3 micro-hardness tester (5 or 30 g load) were used. The results of the X-ray measurements reproduced graphically show that the lattice parameter of < -iron is decreased by phosphorus (Fig.1) and increased by arsenic addition (Fig.2). Deviation from the additivity law, negative in the former and positive in the latter case, was observed. Values of the lattice parameter of the & -phase in alloys annealed in the two-phase region are also given. (The Fe-P alloys were annealed at 400, 650, 835 and 1000°C for 150, 45, 3 and 1.5 hrs respectively, the annealing conditions for the Fe-As alloys being 150, 65,

Card 2/4

SUV/126-6-4-13/34

Effect of Phosphorus and Arsenic on the Lattice Parameter and Hardness of & -Iron

45 and 10 hrs at 400, 500, 650 and 810°C)
From these data solid solubility of P and As in
at various temperatures was determined. The solid
solubility curve of phosphorus in
-iron is shown on
Fig.3, that showing solubility of arsenic in
-iron
is reproduced in another article (Ref.12). Both
phosphorus and arsenic were found to increase hardness
of
-iron. The micro and macro-hardness curves
(graphs a and b) for the Fe P and Fe-As Alloys are
shown on Fig.4 and 5 respectively. The difference
between the values of micro and macro-hardness are
attributed to the fact that the former was determined

dard 3/4

SOV/126-6-4-13/34

. Effect of Phosphorus and Arsenic on the Lattice Parameter and Hardness of & -Iron

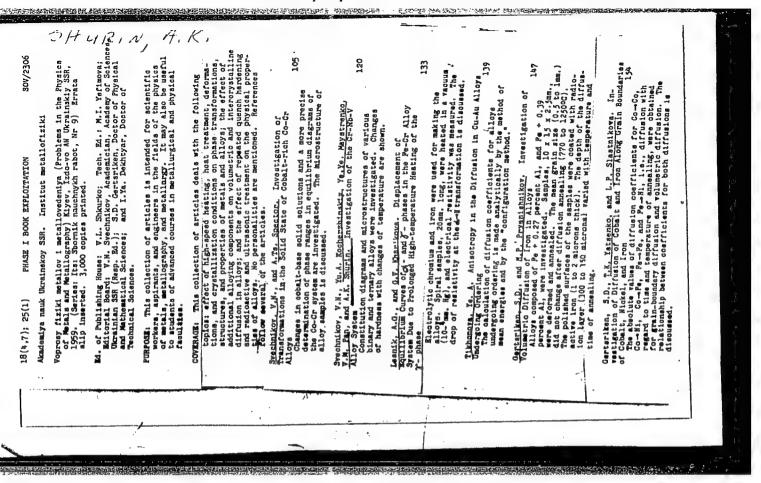
on cold-worked and the latter on annealed specimens. There are 5 figures and 12 references of which 2 are Soviet, 4 English, 3 German, 1 Ukrainian and 2 Norwegian.

ASSOCIATION: Institut Metallofiziki AN USSR

(Institute of Metal Physics, AS UkrSSR)

SUBMITTED: 26th September 1957.

Card 4/4



"APPROVED FOR RELEASE: 08/23/2000

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	30V/3355	Nauchnyy dovet po	<pre>lasisdovaniya po zharoprochnym splavam, t. IV (Studies on Heat-F: aistant Alloys, vol. 4), Moscow, Izd-vo AN SSSR, 1959. 400 p. Errata slip inserted. 2,200 copies printed.</pre>	W. A. Klimov; Tech. Ed.: A. F. Guseva; Bardin, Academician; O. Y. Kurdymnov, ev; Corresponding Member, USSR Academy of I. M. Favlov, and I. F. Zudin, Candidate	FURFOSE: This book is intended for metallurgists concerned with the structural metallurgy of elloys.	ERACE: This is a collection of specialized studies of various problems in the structural metallurgy of heat-resistant alloys Some are concerned with theoretical principles, some with descriptions of new equipment and methods, others with properties temperature.		55ES/408	Jayor, A. P., and I. V. Chemenko, Effect of Plastic Deformation at Low Temperatures on the Hast-esistent Properties of Type 18-8-T1 Austenitic Steal	Savitskir, Te. M., and M. A. Tylkina, Recrystallization of the Refractory Metals Titanium, Marnium, Tantalum, Rhenium, and Tungsten, and Their Alloys	lenko. Br	ton of Pure	udy of the	'/-T., and D. I. Prokof'yey, Constitution Ternary System Chromium-Tungsten-Molybdeina				
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SVECHNIKOV, V.N.; KOCHHRZHINSKIY, Yu.A.; PAN, V.M.; MAYSTREHKO, Ye.Ye.; SHURIN, A.K.

Investigating the chromium - niobium - vanadium system. Issl. po zharopr.splav. 4:248-246 '59. (MIRA 13:5) (Chromium-niobium-vanadium alloys)

SVECHNIKOV, V.N.; KOCHERZHINSKIY, Yu.A.; MAYSTRENKO, Ye.Ye.; PAN, V.M.; SHURIN, A.K.

Investigating the Cr - Nb - V system. Sbor. nauch. rab. Inst. metallofiz. AN URSR no.9:120-132 '59. (MIRA 12:9) (Chromium-niobium-vanadium alloys--Metallography) (Phase rule and equilibrium)

"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001550220020-4

SVECHNIKOV, V.N.; SHURIN, A.K.

Combined effect of manganese and arsenic on transformations in iron-carbon alloys. Sbor. nauch. rab. Inst. metallofiz. AN URSR no.11:53-60.160.

(Iron alloys--Metallography) (Phase rule and equilibrium)

25858 \$/020/61/139/004/018/025 B103/B206

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Svechnikov, V. N., Academician AS UkrSSR, and Shurin, A. K.

TITLE:

AUTHORS:

The phase diagram iron - hafnium

PERIODICAL: Akademiya nauk SSSR. Doklady, v. 139, no. 4, 1961, 895-898

TEXT: The authors investigated the system Fe-Hf and drew up its phase diagram (Fig. 3). The properties of Hf have been investigated insufficiently; the authors were unable to find publications on the insufficiently; the authors were unable to find publications on the could not be produced without large amounts of impurities. Its melting could not be produced without large amounts of impurities. Its melting point has not been established either, nor the temperature of the point has not been established either, nor the temperature of the allotropic transformation existing in Hf. The authors prepared most of the ailoys on the basis of electrolytic iron which was annealed in the ailoys on the basis of electrolytic iron was used for part of the hydrogen and then in vacuo. Carbonyl iron was used for part of the hydrogen and then in vacuo. Carbonyl iron was used for part of the alloys. After purification, the iron contained a maximum of 0.01% C, alloys. After purification, the iron contained a maximum of 0.01% C, alloys. After purification, the iron contained a maximum of 0.01% C, alloys. After purification, the iron contained a maximum of 0.01% C, alloys. After purification, the iron contained a maximum of 0.01% C, alloys. After purification, the iron contained a maximum of 0.01% C, alloys. The apparatus used for the thermal furnace 5 to 6 times in pure argon. The apparatus used for the thermal card 1/6

25858 S/020/61/139/004/018/025 B103/B206

The phase diagram iron - hafnium

THE PROPERTY OF THE PROPERTY O

analysis was described previously (V. N. Sve_nnikov & al., Ref. 12; Mashinostroyeniye, No. 5, 76 (1960)). Tungsten-iridium thermocouples and BP5/20 (VR 5/20) (W + 5% Re — W + 20% Re) were used. The dilatometer is also described in Ref. 12. The phase components of the alloys were determined by X-ray structural analysis. A filtered cobalt Kq-radiation in a cylindrical camera was used. The Curie point of the alloys was determined_with the anisometer by Akulov [Abstracter's note: Anisometer not stated] (methods: V. G. Permyakov & al., Zav. lab., 21, No.6,695 1955)). The authors established that in alloys with less than 45% Hf four transformations take place in the solid state: Two magnetic ones in lpha -iron and in the intermetallide, a third which is linked with the transformation of d-iron into firon, and a fourth which corresponds to the transformation of y-iron into d-iron. The heat of transformation for was determined dilatometrically at a heating rate of up to 0.5 deg/min. The &--> transformation took place in the initial iron (Ccontent up to 0.01%) at a heating rate of 0.2-0.3 deg/min between 894 and 905°C. The heating and cooling dilatograms do not permit the distinction between the transformation $\alpha + \epsilon \rightarrow \gamma + \epsilon$ and $\alpha \rightarrow \gamma$. The greatest solubility of Hf in α -Fe (peritectic point) amounts to 0.2%. It was Card 2/6

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The phase diagram iron - hafnium

determined on the basis of the intersection point of the peritectic horizontal with the extrapolated curve of the α transformation. Characteristic features were a) the temperature increase of the transformation $\alpha + \varepsilon$ with increasing Hf content in alloys, and b) the spreading of the transformation over a considerable temperature interval.

For this reason, the above temperature which amounts to $935^{+}5^{\circ}C$, was determined from the dilatogram of the two-phase alloy $(\alpha + \xi)$, whose composition lies closest to the peritectic point. The transformation

composition lies closest to the partial partial reaction at $1330^{\pm}5^{\circ}C$. The

eutectic alloy contains 2.8% Hf. At 1330° C, the maximum Hf solubility in V-Fe is 1.6%. In alloys with 70 - 99% Hf, two transformations take place in the solid state: 1) magnetic transformation of the intermetallide; the coefficient of thermal expansion is strongly changed here; 2) eutectic

transformation at 1235 $^{\pm}$ 10°C according to the reaction: ℓ + $\alpha_{\rm Hf} \rightleftharpoons \beta_{\rm Hf} + \ell$.

2) is accompanied by a noticeable thermal effect and a considerable volume increase (during heating). From this, the authors conclude that the

Card 3/6

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The phase diagram iron - hafnium

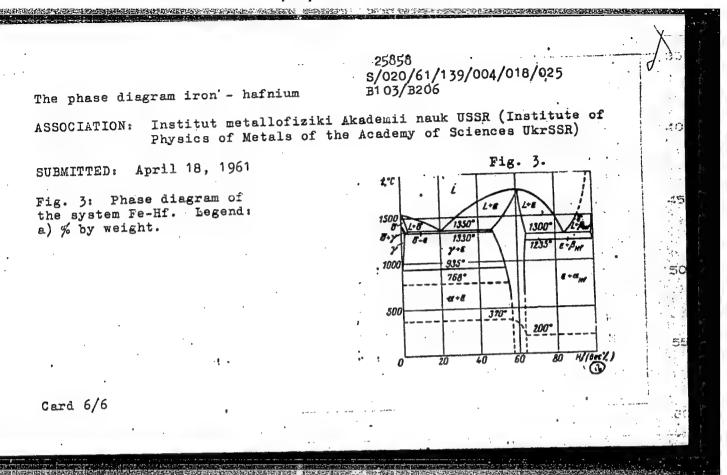
solubility of Fe in Hf is low (certainly below 1%). The fusibility diagram was drawn up only on the basis of a differential thermal analysis. The authors established two eutectic transformations in the system Fe - Hf: A) $L \rightleftharpoons \beta_{Hf} + \mathcal{E}$ at 1300±10°C and B) $L \rightleftharpoons \beta_{+} \mathcal{E}$ at 1350±10°C. The Hf content in the eutectic alloys is 85 and 21.5%, respectively. The crystal lattice of the alloys containing the intermetallide Fe2HF (E-phase) was determined by X-ray structural analysis as being hexagonal and of the MgZn2 type. parameters of this lattice were not the same in alloys of various compositions. From this, the authors conclude that a considerable range of homogeneity of the &-phase exists in some alloys. The boundaries of the mono-phase range were found at 1200°C by direct determination of the hafnium content in the &-phase. I. D. Marchukova (Institut metallurgii AN SSSR - Institute of Metallurgy, AS USSR) made the chemical analysis of this phase by X-ray structural analysis with the PCAW-2 (RSASh-2)apparatus. In the \mathcal{E} -phase of alloy no. 18, 50% Hf were formed, and in no. 21 64%. Since Fe, Hf is ferromagnetic, the Curie point of the ξ -phase was determined Card 4/6

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The phase diagram iron - hafnium

in several alloys, i. e., for various compositions of the &-phase. this way, it was proved that the Curie point drops from 405 to 145°C with an increase of the Hf content in the intermetallide from 50 to 64%. composition of the ?-phase in the alloys was changed owing to long annealing (over 100 hr). The Curie point dropped in alloy .no. 18 from 405 to 377°C, and in no. 21 it increased from 145 to 170°C. boundaries of the mono-phase range of the intermetallide were determined in this way. The hardness of the intermetallide (determined with the Vickers device) is 650 kg/mm², and its melting point is 1810-20°C (for stoichiometric Fe2Hf), which is much higher than that mentioned by R. P. Elliot, W. Rostocker (Trans. Am. Soc. Metals, 50, 617 (1958)). authors thank I. B. Borovskiy for making the spectral analysis in his laboratory. There are 3 figures, 1 table, and 13 references: 2 Sovietbloc and 11 non-Soviet-bloc. The two important references to Englishlanguage publications read as follows: P. Duwes (Ref. 5: J. Appl. Phys., 22, No. 9, 1174 (1951); H. K. Adenstedt (Ref. 2: Trans. Am. Soc. Metals, 44, 949 (1952). The third one see in the body of the abstract. Card 5/6



s/601/62/000/016/016/029 E192/E382

Svechnikov, V.N. and Shurin, A.K.

Constitution diagram of the Fe-C-Hf system AUTHORS:

Intsytut metalo-TITLE:

Akademiya nauk Ukrayins loyi RSR. fyzyky. Sbornik nauchnykh rabot. no. 16. Kiyev, 1962. Voprosy fiziki metallov i metallovedeniya. SOURCE:

In continuation of their earlier work (DAN SSSR, 139, no. 4, 895-898, 1961) on the Fe-Hf system, the authors studied the Fe corner of the Fe-Hf-C constitution diagram, embracing alloys with up to 1.7% C and 20% Hf. High-purity materials were used for the preparation of experimental alloys whose constitution was determined with the aid of thermal and X-ray diffraction analyses, dilatometric measurements and metallographic examination of specimens quenched from various temperatures. Some of the results are reproduced in Fig. 1, showing vertical sections through the Fe-C-Hf system at 0.9, 4, 10 and 20% Hf. The vertical sections at 0.25, 0.55, 1.15. and 1.65% C are shown in Fig. 2. It has been at 0.25, 0.55, 1.15. and 1.65% C are shown in Fig. 2. established that the carbide formed in the Fe-Hf-C alloys Card 1/3

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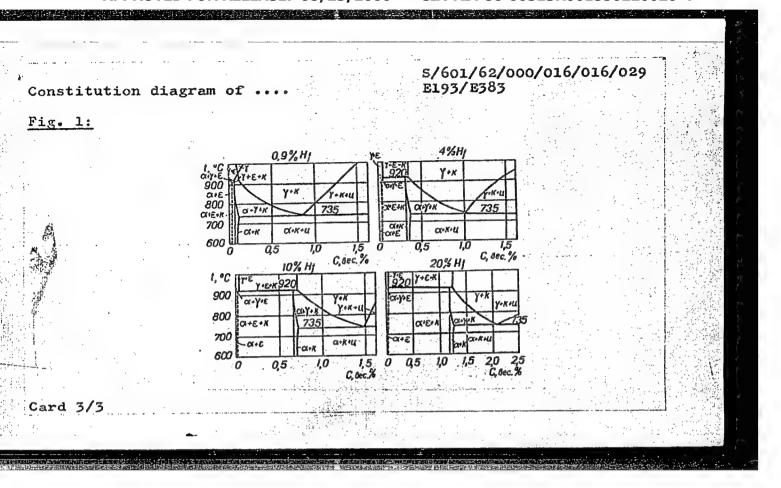
Constitution diagram of ..

formula "HfC"; the regular shape of the HfC crystals indicates that they crystallize out of the melt. In general, Hf sharply decreases the width of the single-phase \gamma-region, considerably increases the C content in the eutectoid and reduces the proportion of pearlite; this constituent is not oberved at C concentrations of less than 0.7% in alloys containing 10% Hf. Practically all the C combines with Hf at low C concentrations and cementite is formed only then the C content is higher than that required to combine all the Hf present. There are two four-phase nonvariant peritectoid equilibria in the concentration range studied:

1) $(\alpha + K \ge (+ \gamma))$ is observed at 735 °C, the four participating phases being $\alpha - \alpha = 0$, $\alpha + \alpha = 0$ (Hfc), $\alpha + \alpha = 0$ is observed at 920 °C, and $\alpha - \alpha = 0$ the other $(\alpha + \alpha = 0)$ is observed at 920 °C, and $\alpha = 0$ the other $(\alpha + \alpha = 0)$ is observed at 920 °C, and $\alpha = 0$ the intermetallic compound Fe Hf. There are 3 figures, denoting the intermetallic compound Fe2Hf. January 26, 1962

SUBMITTED:

Card 2/3



S/601/62/000/016/029/029 E111/E451

AUTHORS: Svechnikov, V.N., Kocherzhinskiy, Yu.A., Shurin, A.K.,

Pan, V.M., Spektor, A.Ts., Kobzenko, G.F., Boyko, Yu.A.

TITLE: Equipment for the physico-chemical investigations on

high-melting chemically active metals

SOURCE: Akademiya nauk Ukrayins'koyi RSR. Instytut metalfyzyky. Sbornik nauchnykh rabot. no.16. Kiev, 1962. Voprosy

fiziki metallov i metallovedeniya. 220-230

TEXT: The following equipment has been developed over several years in the Otdel metallovedeniya (Department of Science of Metals) of Institut metallofiziki AN UkrSSR (Institute of Physics of Metals AS UkrSSR) for studying alloys such as chromium-niobium-vanadium: 1) Arc furnace, including casting facilities, in which evacuation to 10-2 mm is followed by admission of argon to a pressure of 0.2 atm. [Abstracter's note: 10-2 mm is a very poor vacuum and the equipment would not work as described.] The argon is then purified in the furnace by a molten titanium getter. A rotary arrangement enables a clean section of the inspection window to be moved into position without breaking the vacuum.

2) Argon purification plant in which air and moisture are removed Card 1/2

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Equipment for the physico-

S/601/62/000/016/029/029 E111/E451

by calcium chips at 700 to 750°C, through which a 250 litre batch of gas circulates by convection. 3) Installations for annealing specimens in vacuum or argon at temperatures up to 1000°C and up to 2500°C respectively. 4) An installation for differential thermal analysis in an inert medium at temperatures up to 2000°C with novel arrangements for the thermocouple transmitter, thermostat and furnace and taking 0.5 to 1.5 g specimens. Calibration is effected by melting pure metals, the calibration curve then automatically compensating for systematic errors. 5) An inert atmosphere quenching installation (maximum specimen temperature 1400°C). 6) Vertical inert-atmosphere dilatometer and differential dilatometer for temperatures up to 1500°C. There are 10 figures.

SUBMITTED: January 25, 1962

Card 2/2

"APPROVED FOR RELEASE: 08/23/2000 CIA-RDP86-00513R001550220020-4

ACCESSION NR: AT4042839

\$/2601/64/000/018/0170/0174

AUTHOR: Shurin, A. K.; Dmitriyeva, G. P.

大型 (1985年) 1985年 (1986年) 1985年 (1986年) 1985年 (1986年) 1985年 (1986年) 1987年 (1986年) 1987年 (1986年) 1987年 (1986年)

TITLE: Chromium-ruthenium phase diagram

SOURCE: AN UkrSSR. Institut metallofiziki. Sbornik nauchny*kh rabot, no. 18, 1964. Voprosy* fiziki metallov i metallovedeniya (Problems in the physics of metals and physical metallurgy), 170-174

TOPIC TAGS: chromium ruthenium system, chromium ruthenium alloy, alloy phase diagram, alloy composition, alloy solid solution

ABSTRACT: Chromium-ruthenium alloys containing from 2.0 to 58.1 at 7 Ru were prepared by melting electrolytic 99.9% pure Cr and a powder of 99.86% pure metallic Ru in a nonconsumable electrode-arc furnace in an argon atmosphere. The alloys were annealed at 1500C for 25 hr, 1300C for 130 hr in argon, or at 900C for 150 hr and 800C for 100 hr in a vacuum, and water quenched. The phase diagram of the Cr-Ru system (see Fig. 1 of the Enclosure) was plotted on the basis of the differential thermal analysis. The α + β eutectic, where α and β are the solid solutions of Ru in Cr and of Cr in Ru, respectively, is

Card 1/3

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AT4042839 ACCESSION NR:

formed at 1610 \pm 10C and contains 37.5 at% Ru. An intermetallic compound $RuCr_2$ (σ -phase) formed with a peritectoid α + β reaction at about 1580C, has a homogeneity range of ~2 at% Ru (from 35.5 to 37.5 at% Ru). The solubility of Ru in Cr varies from 34 at% at 1600C to 19 at% at 800C. The solubility of Cr in Ru varies from 52.5 at% at 1600C to 46 at% at 900C. Alloys with 20.2 to 31.4 at% Ru annealed at 800 and 900C contain a RuCr3 compound (cubic lattice of the $\beta-W$ type) and a RuCr4 compound, the latter probably formed from the supercooled $\alpha-$ phase containing 18-20 at% Ru. With the decomposition of the RuCr3 compound at 780C into a mixture of the α and σ phases the alloy expands. The hardness of the $\alpha-phase$ increases from 1519 Mn/m^2 for pure Cr to 8820 Mn/m^2 for the maximum saturated solid solution. The σ -phase has a hardness of 10,388 Mn/m^2 , and the maximum saturated β -phase has a hardness of 2040 Mn/m^2 . Orig. art. has: 5 figures and 1 table.

Institut metallofiziki AN UkrSSR (Institute of Physics of Metals, ASSOCIATION:

AN UKrSSR) SUBMITTED:

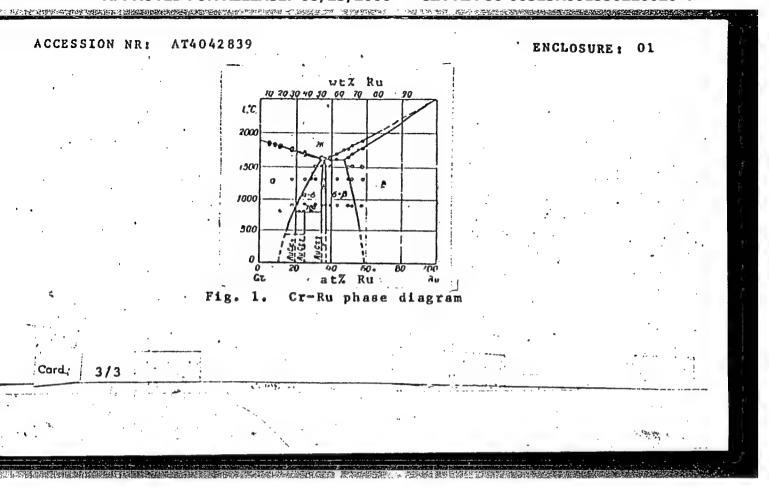
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NO REF SOV: 003 OTHER: 002



\$/2601/64/000/018/0175/0177

ACCESSION NR: AT4042840

AUTHOR: Shurin, A. K.; Dmitriyeva, G. P.

TITLE: Niobium-cobalt phase diagram

SOURCE: AN UkrSSR. Institut metallofiziki. Sbornik nauchny*kh rabot, no. 18, 1964. Voprosy* fiziki metallov i metallovedeniya (Problems in the physics of metals and physical metallurgy), 175-177

TOPIC TAGS: niobium cobalt system, niobium cobalt phase diagram

ABSTRACT: Thirty niobium-cobalt alloys containing from 14.4 to 99.2 at 7 Nb were prepared by melting electrolytic cobalt and 99.4% pure niobium in a nonconsumable tungsten electrode-arc furnace in an argon atmosphere. The alloys were annealed in either argon or in vacuum at a temperature ranging from 1300 to 700C for periods varying from 50 to 435 hours. The phase composition was determined by metallographic and x-ray structural analyses, and a phase diagram of the Nb-Co system, based on data from the differential thermal analysis, was plotted (see Fig. 1 of the Enclosure). The intermetallic compound NbCo3 of the α + NbCo3 eutectic at 700—1000C is homogeneous at an Nb concentration of 28 to 33 at% and has a melting cord 1/3

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ACCESSION NR: AT4042840

temperature of 1540 \pm 10C. The alloys annealed at 700, 1000, or 1100C had only the α and NbCo₂ phases. The intermetallic compound β is formed during crystallization at 1440 \pm 10C and decomposed with the eutectoid reaction $\beta \neq$ NbCo₂+ NbCo at 1225 \pm 20C. Decomposition of the β -phase at 1200C and its formation from the NbCo₂ + NbCo mixture can be achieved with 50-hr annealing. The NbCo intermetallic compound is homogeneous in the 700—1250C range at a Nb content of 45 to 54 at%. The solubility of Co in Nb, determined at 1250C, was about 1.57 at% (1 wt%). Two-phase (NbCo + γ) alloys contained no other phases. Orig. art. has: 2 figures and 1 table.

ASSOCIATION: Institut metallofiziki AN UkrSSR (Institute of Physics of Metals,

AN UkrSSR)

SUBMITTED: 20Feb63

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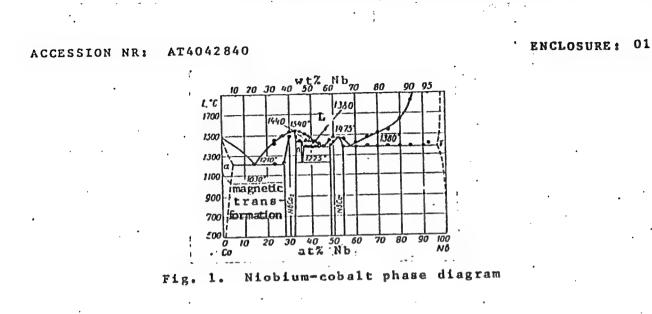
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OTHER: 006

Card 2/3.



ACCESSION NR: AP4037443

s/0021/64/000/005/0600/0603

AUTHOR: Kuz'ma, Yu. B.; Shurin, A. K.; Dmy*triyeva, G. P. (Dmitriyeva, G. P.); Glady*shevs'ky*y, Ye. I. (Glady*shevskiy, Ye. I.)

TITLE: Crystal structure of the beta-phase of the niobium-cobalt system and the solubility of silicon in it

SOURCE: AN UKERSR. Dopovidi, no. 5, 1964, 600-603

TOPIC TAGS: niobium-cobolt system, beta-phase, beta-phase stabilization, x-ray analysis, space group D sup 4 sub 6h, space group P6 sub 3/mmc, MgZn sub 2 structure, niobium-cobalt-silicon system

ABSTRACT: X-ray analysis was used to establish that the beta-phase of the niobium-cobalt system, existing over the temperature interval ll: 10-1225C, has the MgZr₂ structure (space group P6₃/mmc -- D6h) with lattice parameters a = (1.83h \pm 0.002) A, c = (7.853 \pm 0.00h) A, c/a = 1.62h for the alloy containing 35.1 atomic % Nb. The beta-phase had been studied earlier by two of the authors,

Card 1/2

ACCESSION NR: AP4037443

A. K. Sh. and G. P. D. (Voprosy* fiziki metallov i metallovedeniya, no. 18, 1963, p. 175). The other authors, Yu. B. K. and Y. I. G. had previously established the existence of the composition Nb₂Co₃Si at 800°C (Mg₂Cu₃Si structure) in the ternary Nb-Co-Si system. The purpose of this study was to see whether this ternary alloy is a solid solution based on the beta-phase. The results showed the answer to be in the affirmative. They also indicated that the beta-phase of the Nb-Co can dissolve up to 25% atomic % Si, and that the addition of Si can stabilize the beta phase down to 800 C. Orig. art. has one table and one figure.

'ASSOCIATION: L'vivs'ky*y derzhavny*y universy*tet, Insty*tut metalofizy*ky*
AN UkrSSR (L'vov State University, Institute of Metal Physics, AN UkrSSR)

SUBMITTED: 03May63

DATE ACQ: 03Jun64

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SUB CODE: MM,SS

NO REF SOV: OOL

OTHER: 004

Card 2/2

EWT(m)/EPF(n)-2/EPR/EPA(bb)-2/EWP(b) Ps-4/Pu-4 ASD(f)-2/ASD(m)-3 ACCESSION NR: AT4046825 JD/JG/MLK \$/0000/64/000/000/0104/0107 AUTHOR: Svechnikov, V. N.; Shurin, A. K.; Dmitriyeva, G. P. Investigation of alloys of the Nb-NbCr2-NbAl3 system TITLE: 21 ท SOURCE: AN SSSR. Nauchny*y sovet po probleme zharoprochny*kh splavov. Issledovaniya staley i splavov (Studies on steels and alloys). Moscow. Izd-vo Nauka, 1964, 104-107 TOPIC TAGS: niobium base alloy, niobium chromium aluminum system. niobium aluminum intermetallic compound, niobium chromium intermetallic compound, intermetallic compound niobium alloy, alloy hot hardness alloy oxidation rate ABSTRACT A Arc-melted Nb-Al, Nb-Cr, and Nb-Cr-Al alloys were annealed at 1500C for 17-30 hr or at 1200C for 105 hr in an argon atmosphere, and tested for hot hardness in a vacuum at temperatures up to 900C and for oxidation in air at 1100C. At all test temperatures the hardness of Nb-Al alloys increased gradually with the addition of up to 5% Al and increased sharply with further increases in Al content. An almost linear increase in hardness was alloy containing up to 17% Cr. The observed in Nb-Cr

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ACCESSION NR: AT4046825

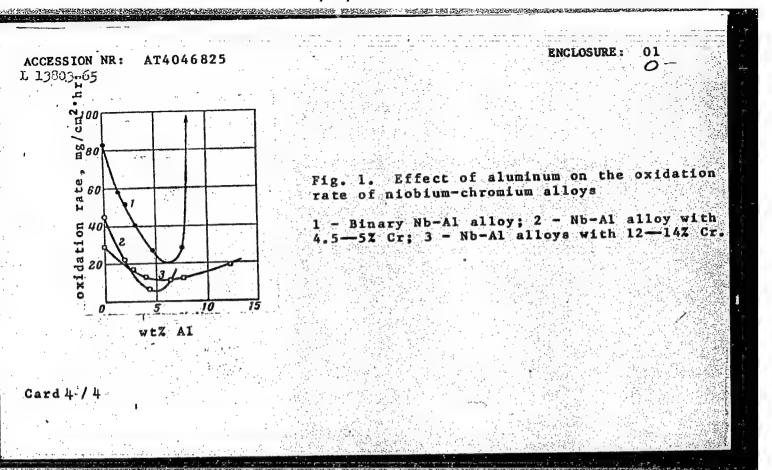
Nb-Cr-Al alloys containing more than 50% intermetallic compounds in the structure also increases appreciably. The oxidation rate of unalloyed Nb and Cr was 83 and 0.59 mg/cm²·hr, respectively. Chromium additions (up to 8-10%) decrease the oxidation rate of Nb-Cr alloys more sharply than do subsequent additions. The exidation rate in these alloys decreases until NbCr₂ starts forming. Fig. 1 of the Enclosure shows the effect of Al on the oxidation rate of Nb-Cr alloys. The oxidation rate of an Nb-Al₃ compound (0.84 mg/cm²·hr) is comparable to that of Cr. Alloys of the NbCr₂-NbAl₃ section have good oxidation resistance. Thus the Nb-Cr-Al system has several Nb-base alloys whose oxidation rate is below that of adjacent alloys containing more or less Al but with the same Cr content. Alloys containing 5-6 wt% Al or 4-7 wt% Cr have their lowest (5 mg/cm²·hr) oxidation rate at 1100C, i.e., 16 times below that of an unalloyed Nb. These alloys have a hardness of 400-470 and 290-350 kg/mm² at 20 and 900C, respectively. Orig. art. has: 3 figures and 1 table.

ASSOCIATION: none

Card 2/4

L 13803-65
ACCESSION NR: AT4046825

SUBMITTED: 16Jun64 ENCL: 01 SUB CODE: MM
NO REF SOV: 011 OTHER: 008 ATD PRESS: 3131



SHURIN, A.K.; DMITRIYEVA, G.P.

Constitutional diagram of the chromium-ruthenium alloy. Sbor. nauch. rab. Inst. metallofiz. AN URSR no.18:170-174 64 (MIRA 17:8)

Gonstitutional diagram of the niobium-cobalt alloy. Ibid.:175-177.

EPF(n)-2/EWP(k)/EWP(z)/EWA(c)/EWT(m)/EWP(b)/T/EWA(d)/EWP(e)/EWP(w)/IJP(c) JD/JG EWP(t) Pf-4/Pu-4 S/2601/64/000/019/0206/0211 ACCESSION NR: AT5005125 Svechnikov, V. N. (Academician AN UkrSSR); Shurin, A. K.; Dmitriyeva,

AUTHOR:

G. P.

Phase diagram of the Hf-Ta system in the solid state TITLE:

B+1

SOURCE: AN UkrSSR. Institut metallofiziki. Sbornik nauchnykh trudov, no. 19, 1964. Voprosy fiziki metallov i metallovedeniya (Problems in the physics of metals and physical metallurgy), 206-211

TOPIC TAGS: hafnium, tantalum, hafnium tantalum system, hafnium alloy, alloy microstructure, alloy hardness, tantalum containing alloy composition

ABSTRACT: Sixteen binary Hf-Ta alloys, containing from 2.5 tg 95 wt% Ta, arc melted in a tungsten electrode furnace in an argon atmosphere were investigated. Thermal analysis revealed no signs of melting at temperatures up to 1800C. However, an endothermal solid-state transformation was observed at 1000-1050C. A supersaturated β_2 -phase (a solid solution of Ta in Hf), stable at room temperature, was obtained by quenching alloy powders from 1300 and 1500C. The solubility of Hf in Ta was found to be 6, 8.8, 16, and 33 wt% at 900, 1100, 1300, and 1500C, respectively. The x-ray diffraction patterns of Ta and an alloy with 80% Ta showed only the β2-

Card 1/3>

L 39919-65

ACCESSION NR: AT5005125

phase lines (see Fig. 1 of the Enclosure). The diffraction pattern of Hf showed only the α-phase lines. However, the patterns of all alloys with up to 70% Ta contained no a-phase lines but those of a'-phase, a supersaturated solid solution of Ta in α-Hf. The maximum solubility of Ta in α-Hf was about 5 wt%. Alloys containing 10-20% Ta quenched from 1100, 1300, 1500, and 1700C contained an ω -phase. The hardness of α + β_2 alloys annealed at 900C changes linearly, but with annealing at 1100C, it sharply rises at the Ta concentrations at which the solid w-phase is formed. Orig. art. has: 5 figures and 1 table.

ASSOCIATION: Institut metalloffiziki AN UkrSSR (Institute of the Physics of Metals,

AN UkrSSR)

SUBMITTED: 26Jun63

ENCL: 01

SUB CODE:

NO REF SOV: 002 OTHER: 016

ATD PRESS: 3191

Card 2/3

SHURIN, A.K.

Differential highly sensitive dilatometer with heating in an inert atmosphere. Sbor. nauch. rab. Inst. metalloriz. AN URSR no.18:222-225 *64 (MIRA 17:8)

L 41180_65 EWT(m)/T/EWP(t)/EWP(b)/EWA(c) P1_4 IJP(c) RWH/JD/JG
ACCESSION NR: AP4046381 S/0020/64/158/003/0668/0670 23

**ACCESSION NR: AP4046381 B Dmitriveya G. P.: Kobzenko.

AUTHOR: Svechnikov, V.N. (Academician AN UkrSSR); Dmitriyeva, G.P.; Kobzenko, G.F.; Shurin, A.K.

TITLE: Diagram of phase equilibria of the chromium-osmium system

SOURCE: AN SSSR. Doklady*, v, 158, no. 3, 1964; 668-670

TOPIC TAGS: phase equilibrium chromium osmium system, eutectic alloy, eutectoid reaction, chromium alloy, osmium alloy

ABSTRACT: Alloys of Cr and Os were made in an arc furnace with a Cu, water-cooled hearth, in an argon medium. To eliminate possible dendrite liquation, the alloys were then subjected to homogenizing annealing at 1700C for 55 hrs. The phase equilibria diagram shown in Figure 1 of the Enclosure is of the eutectic type. The temperature of the eutectic reaction ($L \rightleftharpoons A + A$) is $1840 \pm 10C$. The eutectic alloy contains 33 at.%(64 wt.%) Os. The σ -phase(Cr₂Os) forms after crystallization of the eutectic in accordance with the peritectoid reaction $A + A \rightarrow \sigma$ at 1670 \pm 15C. Upon further cooling of the alloys, there is a further decomposition of the σ -phase in accordance with the eutectoid reaction $\sigma \rightarrow Cr_3Os + A$ at 975 \pm 25C. At 1670C, the σ -phase forms and decomposes under conditions of continuous heating and cooling. The compound Cr₃Os is obtained in accordance

Card 1/2 2

L 41180-65 ACCESSION NR: AP4046381

2

with the peritectoid reaction $3+\pi\to Cr_3Os$ after prolonged annealing at 1540 \pm 40C. Hence, cast alloys of Cr_3Os consist of a solid solution of Os in chromium (d-phase). This same alloy, annealed at 1700C, has the structure $3+\pi$, and after annealing at 1500C and below, Cr_3Os forms. The Cr_3Os has a certain region of homogeneity. An increase in Os from 0 to 25 at.% increases microscopic hardness (from 150 to 600 kg/mm²) and the crystal lattice (from 2885 Å to 2930 Å) of chromium. Cr_3Os has a hardness of 600 kg/mm², while the hardness of the σ -phase, depending on the composition, ranges from 1800 to 2000 kg/mm², and the hardness of the saturated β -phase is about 800 kg/mm². Orig. art. has: 3 figures and 1 table.

ASSOCIATION: Institut metallofiziki Akademii Nauk UkrSSR (Institute of the Physics of Metals, Academy of Sciences, Ukr SSR)

SUBMITTED: 25Mar64

ENCL: 01

SUB CODE: MM

NO REF SOV: 000

OTHER: 003

Card 2/3

L. 41560-65 EWT(m)/EWP(w)/EPF(c)/EPF(n)-2/EWP(v)/T/EWP(t)/EWP(b)/EWA(c) 39

AUTHOR: Alfintsova, R. A.; Dmitriyeva, G. P.; Korobeynikova, V. G.;
Pan, V. M.; Shurin, A. K.; Svechnikov, V. N. (Academician An UkrSSR)

TITLE: Investigation of chromium-iron-molybdenum and chromium-iron-tungsten alloys

SOURCE: AN UkrSSR. Institut metallofiziki. Sbornik nauchnykh trudov, no. 20, 1964. Voprosy fiziki metallov i metallovedeniya (Problems in the physics of metals and physical metallurgy), 108-124

TOPIC TAGS: chromium alloy, iron containing alloy, molybdenum containing alloy, tungsten containing alloy, alloy structure alloy hot hardness, alloy oxidation resistance

ABSTRACT: The following alloys have been investigated to determine which ternary Cr-Fe-Mo or Cr-Fe-W alloy would provide the optimum combination of the heat resistance of Mo or W and the ductility of Cr: binary chromium-iron alloys containing 45-90% Cr, chromium-molybdenum alloys containing 10-30% W, and ternary alloys containing up to 55% Fe and Cord 1/3

L 41560-65
ACCESSION NR: AT5008875

up to 30% Mo or W. In Cr-Fe-Mo alloys containing 45-50% Cr, additions of up to 6% Mo do not improve hot hardness or oxidation resistance. Increasing Mo content leads to the formation of a brittle o-phase which has a very low oxidation resistance in air at 1100C and lowers the oxidation resistance of the ternary Cr-Fe-Mo alloys in direct proportion to its content in the alloys. In Cr-Fe-W alloys, the single phase o-region extends to about 32% W, but it tapers off at about 1275C. At high temperature (1450C), the single-phase region of c-solid solution with a b.c.c. lattice increases substantially, so that all the investigated alloys, except for an alloy containing 40% Fe and 30% W, became single-phase alloys at a more or less high temperature. A single-phase structure and a satisfactory ductility

is readily preserved in all but three of these alloys by oil quenching from 1450C. Tungsten additions increase somewhat the melting temperature of Cr-Fe alloys, e.g., 30% W increases the solidus temperature by 100 and 150C in alloys with 40 and 50% Fe, respectively. Tungsten also increases the hardness of Cr-Fe-W alloys at both room and high temperature and does not impair their oxidation resistance. Orig. art. has: 8 figures and 3 tables. [MS]

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ACC NR: AT6010590

SOURCE CODE: UR/0000/65/000/000/0159/0162

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AUTHOR: Svechnikov, V.N. (Academician AN UkrSSR); Shurin, A.K.; Dmitriyeva, G.P.

ORG: Institute of Metal Physics, AN UkrSSR (Institut metallofiziki AN UkrSSR)

32 B+1

TITLE: The hafnium-chromium phase diagram 4

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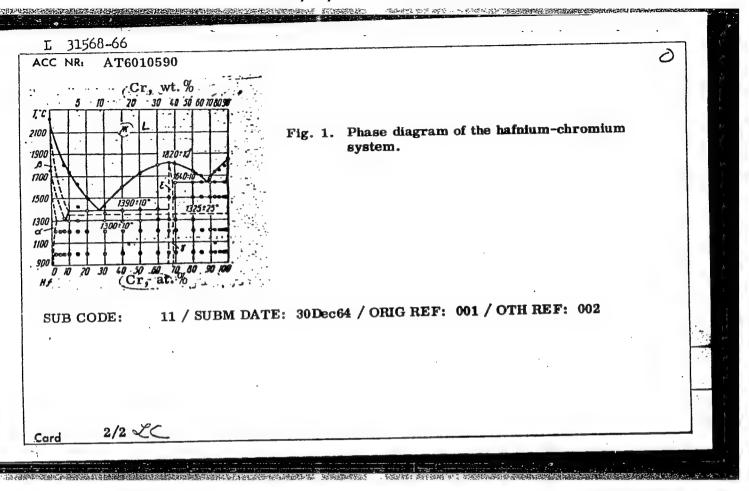
SOURCE: AN UkrSSR. Fazovyye prevrashcheniya v metallakh i splavakh (Phase iransformations in metals and alloys). Kiev. Naukova dumka. 1965, 159-162

TOPIC TAGS: hafnium alloy, chromium alloy, alloy phase diagram

ABSTRACT: The Hf-Cr system was investigated by metallographic and x-ray phase analysis of Hf-Cr alloys prepared in an arc furnace and subjected to various annealing treatments. Differential thermal analysis was used to determine the temperatures of the start and end of fusion. The phase equilibrium diagram obtained is shown in Fig. 1. The eutectoid equilibrium $\beta \rightleftharpoons \alpha + \gamma$ was observed in Hf-rich alloys at $1300 \pm 10 \text{C}$. Metallographic analysis of alloys annealed at 1500, 1200, and 1000C showed that alloys containing from 70 to 99.9 at.% Cr consist of two phases, and those with 99.95 and 99.97 at.% Cr have a single-phase structure. Thus, the solubility of hafnium in chromium was found to be 0.05-0.1 at.%; it remains practically unchanged as the temperature is lowered. Metallographic analysis of specimens annealed at 1200 and 1000C established that the solubility of chromium in hafnium does not exceed 2 at.% at these temperatures. The temperature of the polymorphous transformation in the compound HfCr₂ was found to be 1325 \pm 20C. Orig. art. has: 1 figure and 1 table.

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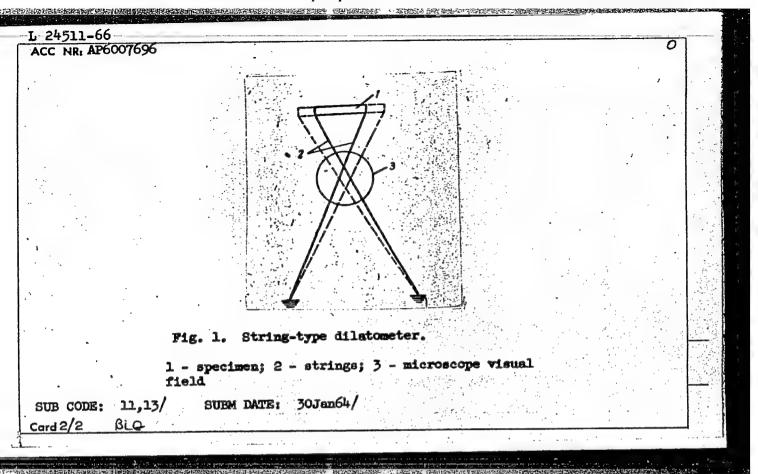


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L 24511-66 EWT(d)/I ACC NRAP6007696	EWP(h)/EWP(l) SOURCE	CODE: UR/0413/66/000/003/0075	
AUTHOR: Krakovetskiy	-Kocherzhinski y, Yu. A.	; Shurin, A. K.	, I'll
ORG: none TITLE: String-type d	ilatometer, Class 42,	No. 178530 [Announced by the <u>I</u> allofiziki AN USSR)]	nstitute
SOURCE: Izobreteniya	, promyshlennyye obrazt	sy, tovarnyye znaki, no. 3, 19	66, 75
ABSTRACT: An Author measuring the elongat	Certificate has been in	sued for a string-type dilatom ery small dimensions (WTO improdure, the indicator strings are the within the visual field of re-	eter for ove the positioned
crosswise, and their microscope (see Fig.	boir of jurelaccion r	within the visual field of re	[m]
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Name: SHURIN, A. T.

Dissertation: Benthonic fauna of the Gulf of Riga

Degree: Cand Biol Sci

Acad Sci USSR, Zoological Inst Acad Sci USSR, Academic

Council

Defense Date, Place: 1956, Leningrad

Source: Knizhnaya Letopis', No 51, 1956

SHURIN, A.T., kand. biologicheskikh nauk

Bottom fauna of the Gulf of Riga and factors governing its distribution. Trudy VNIRO 42:37-60 '60. (MIRA 13:9) (Riga, Gulf of-Benthos)

SHIRI, L.

Shurin, L. - "Engines", (On the history of technology), Illustrated by N. Yanitskiy, Ananiya--sila, 1949, No. 3, p. 9-14.

SO: L-4110, 17 July 53, (Letopis 'Zhurnal 'nykh Statey, 4. 19, 1949).

SHURIN, P.S., inzh.

Specialization and cooperation in the electrical industry. Vest. elektroprom. 31 no.3:75-76 Mr 160. (MIRA 13:6)

 Glavnyy tekhnolog Novosibirskogo Giproenergoproma. (Electric engineering)

SHURIN, S.N.

Technical progress in sement plants in Novorossiysk. TSement 27 no.5:7-9 S-0 '61. (MTRA 14:12)

1. Pervyy sekretar! Novorossiyskogo Gorkoma Kommunisticheskoy partii Sovetskogo Soyu a.

(Novorossiyek.—Cement industries)

USSR / Human and Animal Physiology. Blood Circulation.

T

Abs Jour

: Ref Zhur - Biologiya, No 1, 1959, No. 3398

Author

: Shurin, S. P.

Inst

: Novosibirsk Medical Institute

Title

: Some Data on the Role of the Central Nervous System in the Process of Formation and Development of Experi-

mental Endomyocarditides. Preliminary Report

Orig Pub

: Tr. Novosibirskogo med. in-ta, 1957, 27, 74-94

Abstract

: Studies were conducted on 104 normal non-sensitized rabbits. When the carotid artery of 10 rabbits was infected with blood taken from a patient with rheumatism and processed with hyaluronidase, inflammatory changes appeared in the cardiac muscle more often than when other modes of administration were used. Upon introduction into the carotid artery of a hyaluronidase-treated extract from the valves and myocardium of a

Card 1/3

ZALRSSKIY, G.D., prof., VOROB'YEVA, N.N., prof., PIROGOVA, O.I., SHURIN, S.P., KAZHACHEYEV, V.P., YAVOROVSKAYA, B.Ye., FEDOROV, A.I., MOSOLOV, A.N.

Specific agent inducing rheumatic fever. Report No.1: Some data on a filtrable virus isolated in rheumatic fever. Terap. arkh. 30 no.5:3-15 My '58 (MIRA 11:6)

1. Iz Novosibirskogo meditsinskogo instituta. (RHEUMATIC FEVER, microbiology, isolation & infect. of animals with specific virus (Rus)) (VIRUSES, isolation & infect. of animals with specific rheum. virus (Rus))

SHURIN, S.P.

Some data on the role of the lymphogenic spread of infection in the pathogenesis of experimental endomyocarditis. Terap.arkh. 30 no.5:15-21 My '58 (MIRA 11:6)

l. Iz kafedry fakul'tetskoy terapii (zav. - prof. G.D. Zalesskiy)
Novosibirskogo meditsinskogo instituta.

(RHEUMATIC HEART DISEASE, experimental
lymphatic spreading of infect. (Rus))

(LYMPHATIC SYSTEM, physiology,
spreading of infect. in exper. rheum. heart dis.

(Rus))

SHURIN, S. P., CAND MED SCI, "EXPERIMENTAL INVESTIGATIONS ON REPRODUCTION OF RHEUMATIC CARDITIS IN RABBITS." NOVOSI-BIRSK, 1959. (NOVOSIBIRSK STATE MED INST). (KL-DV, 11-61, 231).

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DREYZIN, R.S., YAVOROVSKAYA, V.Ye., BALANDINA, A.M., SHURIN, S.P., VORORIYEVA, N.N., MOSOLOV, A.N., ZALESSKIY, GIDI, ZHDANOV, V.M.

Group of new virus strains, the so-called R virus. Vop. virus. 6 no.5:521-532 S-0 '61. (MIRA 15:1)

1. Institut virusologii imeni D.I.Ivanovskogo AMN SSSR, Moskva i Novosibirskiy meditsinskiy institut, Novosibirsk. (VIRUSES)

2000年,1900年,1900年的1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1900年,1

SHURIN, S.P.; YAVORSKAYA, V.Ye.; LOZOVOY, V.P.

Detection of the virus isolated from patients with rheumatic fever in a culture of fibroblasts with the aid of labelled fluorescent serum. Vop.virus. 7 no.3:273-276 My-Je '61. (MIRA 14:7)

l. Iz kafedry fakul'tetskoy terapii i revmatologicheskoy laboratorii Novosibirśkogo meditsinskogo instituta. (RHEUMATIC FEVER) (VIRUSES)